An Integrated Renewable Energy/ Fuel Cell Power Generation System - a preliminary study -

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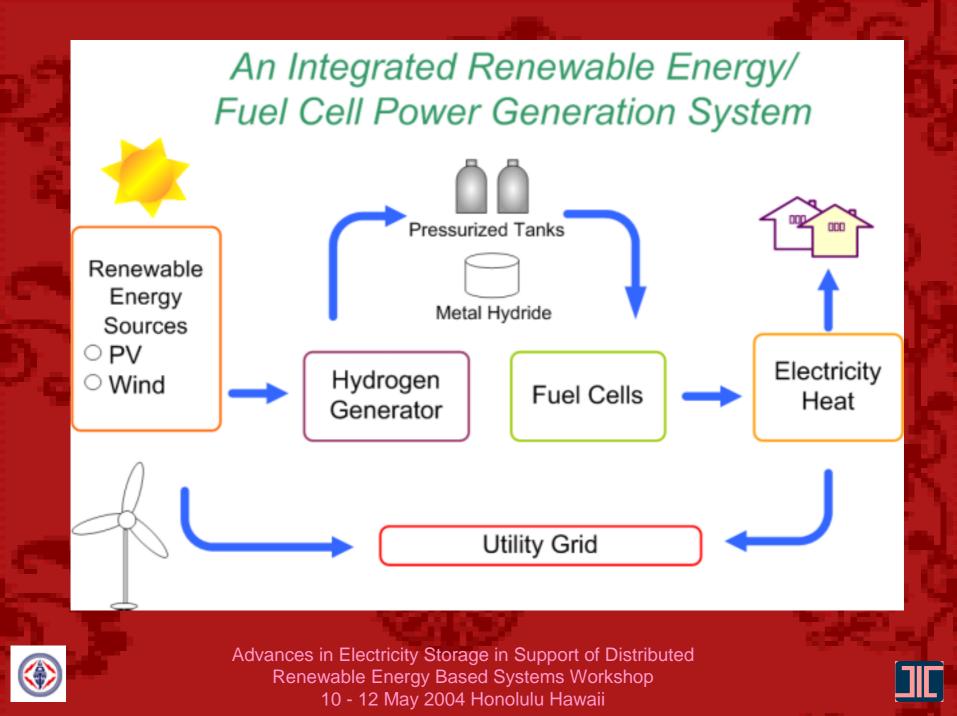


Comparison of Different Electricity Storage Technologies

MODE	ENERGY FORM	ADVANTAGE	DISADVANTAGE
Pumped Hydro	Potential	Matured technology, quick response	Location
Compressed Air	Internal	Matured technology	Location, need extra auxiliary power
Superconductor	Magnetic	Efficiency	Technology under research
Battery	Chemical	No limitation on location, quick response	Cost, life, weight, safety
Regenerative Fuel Cells	Chemical	Efficiency, environmentally friendly	Developing technology, cost, capacity limitation







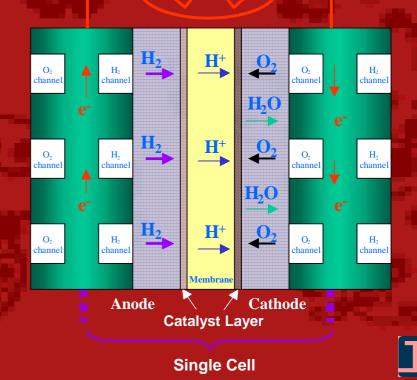
What Is a Fuel Cell ?

	Combining hydrogen and o	xygen to generate electricity and heat
Basic	Anode	Cathode
Reactions	$H_2 \rightarrow 2H^+ + 2e^-$	$O_2 + 4H^+ + 4e^- \rightarrow 2H_2O$

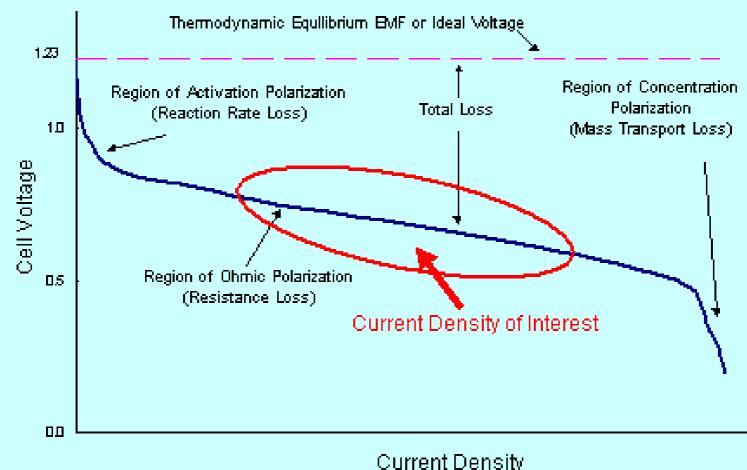
- A device that converts the chemical energy of fuels directly into electricity and heat via electrochemical reactions
- Similar to a primary battery except that the source of energy is continuously fed from outside
- Individual fuel cells are assembled into modules (stacks) to provide a practical voltage and power output

Membrane Electrode Assembly (MEA)

Load













Reactant Gas Flow Rates for Fuel Cells Generating 1 kW Electricity

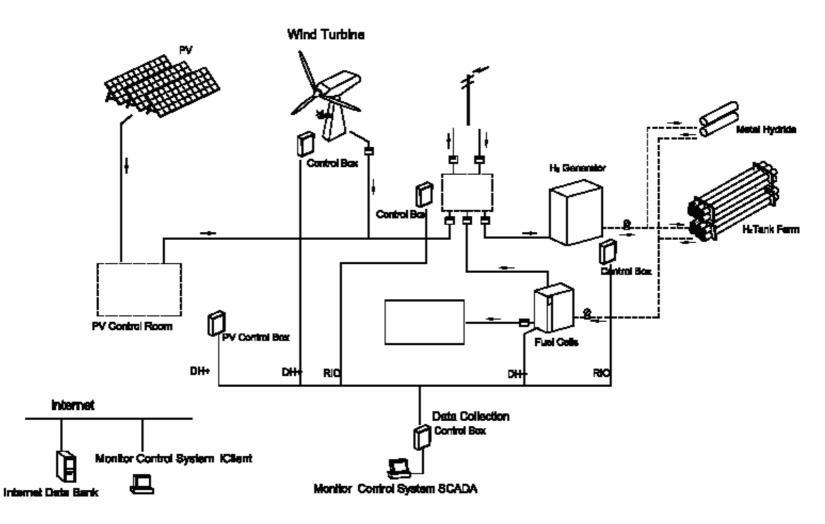
	Hydrogen			Oxygen		Air		
Voltage per cell (volts)	1.0 St (slpm) [#]	Fuel * Economy (%)	1.2 St (slpm) [#]	Fuel * Economy (%)	1.0 St (slpm) [#]	2.0 St (slpm) [#]	1.0 St (slpm) [#]	2.0St (slpm) [#]
1.0	6.964	79.75	8.356	66.47	3.482	6.694	16.580	33.161
0.9	7.737	71.78	9.285	59.82	3.869	7.737	18.423	36.846
0.8	8.705	63.80	10.446	53.17	4.352	8.705	20.725	41.451
0.7	9.948	55.83	11.938	46.52	4.974	9.948	23.686	47.372
0.6	11.606	47.85	13.927	39.88	5.803	11.606	27.633	55.268
0.5	13.927	39.88	16.713	33.23	6.964	13.927	33.160	66.321
0.4	17.409	31.90	20.891	26.59	8.705	17.409	41.451	82.901
0.3	23.212	23.93	27.855	19.94	11.606	23.212	55.268	110.535

based on LHV of hydrogen (57.7979 kcal/mole) standard liters per minutes (0 °C and 1 atm)





SYSTEM CONFIGURATION





Major Components 1 Renewable Energy Sources photovoltaic power generation (1)wind power generation 2 **2** Hydrogen Production and Storage PEM vs. alkaline electrolysis (1)pressurized tanks vs. metal hydride 2 **3** Fuel Cells Power Generation System **4** Load Display 5 Monitoring and Control System

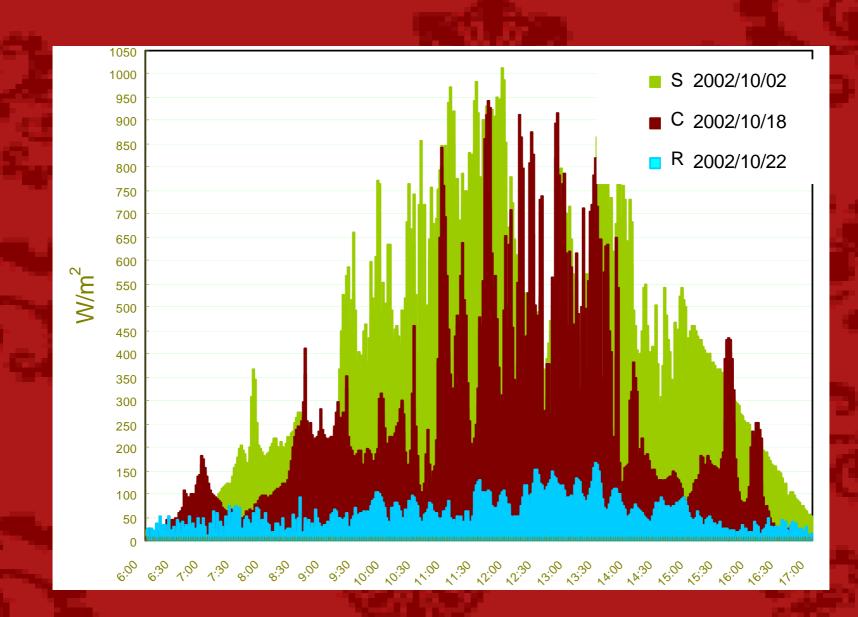




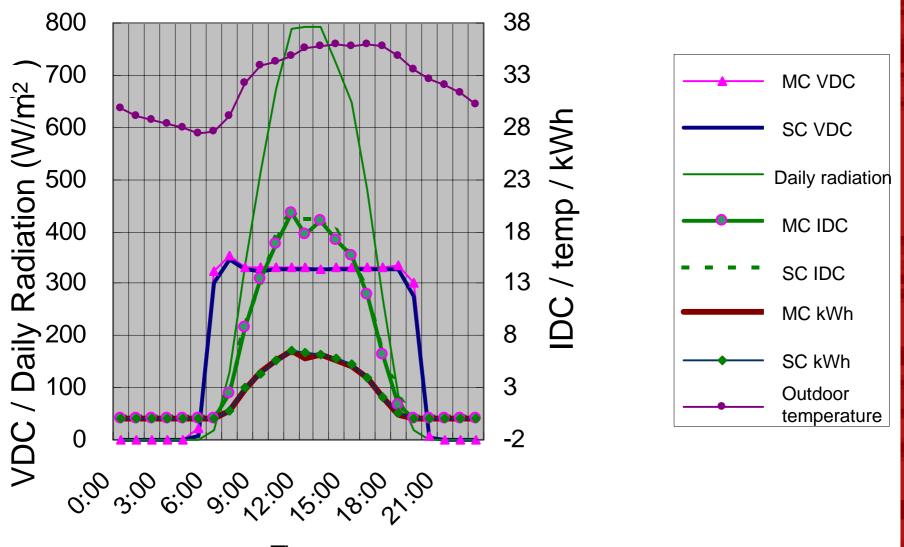
Photovoltaic Power Generation







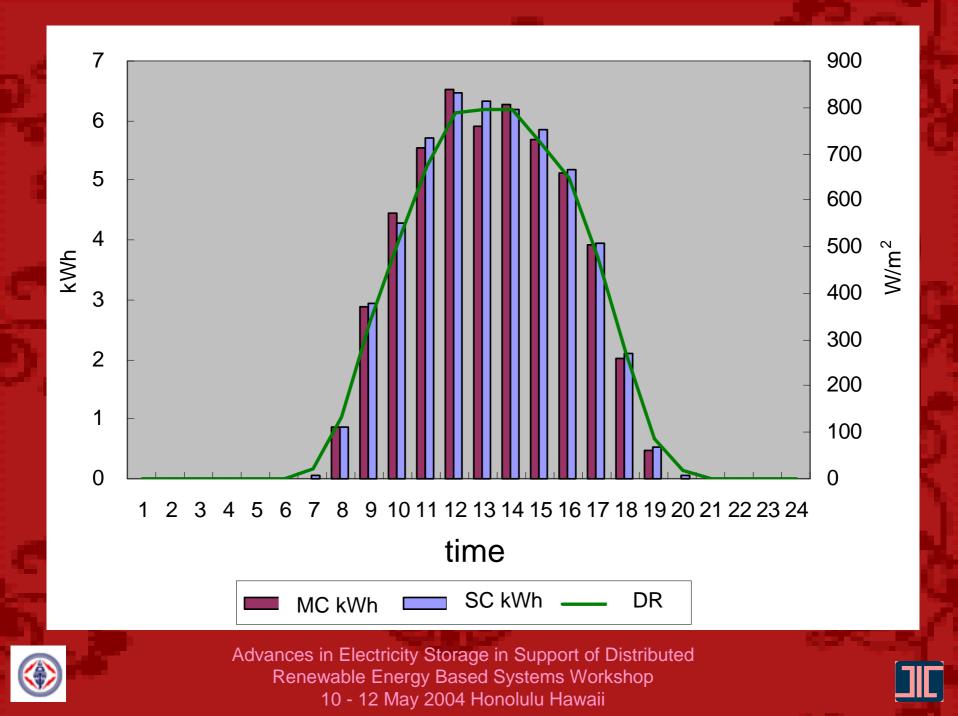


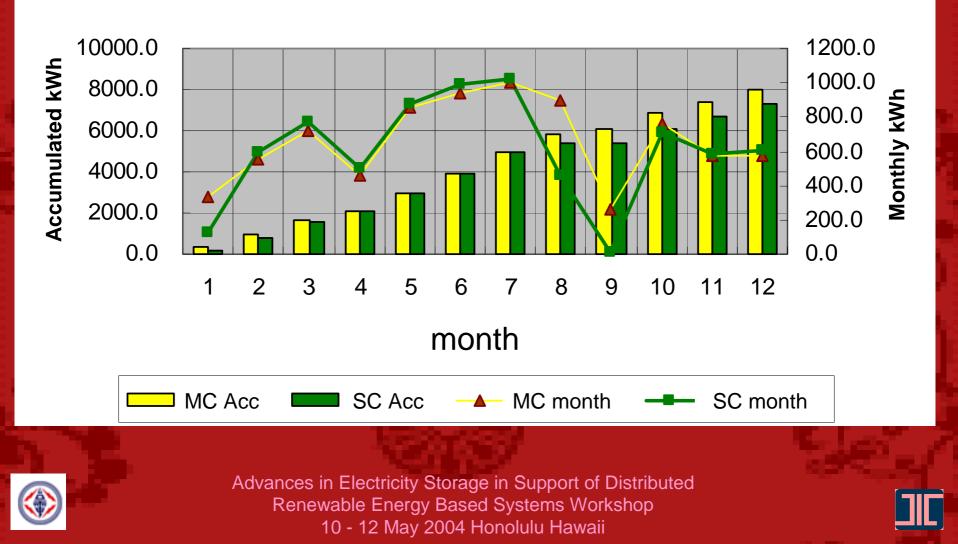


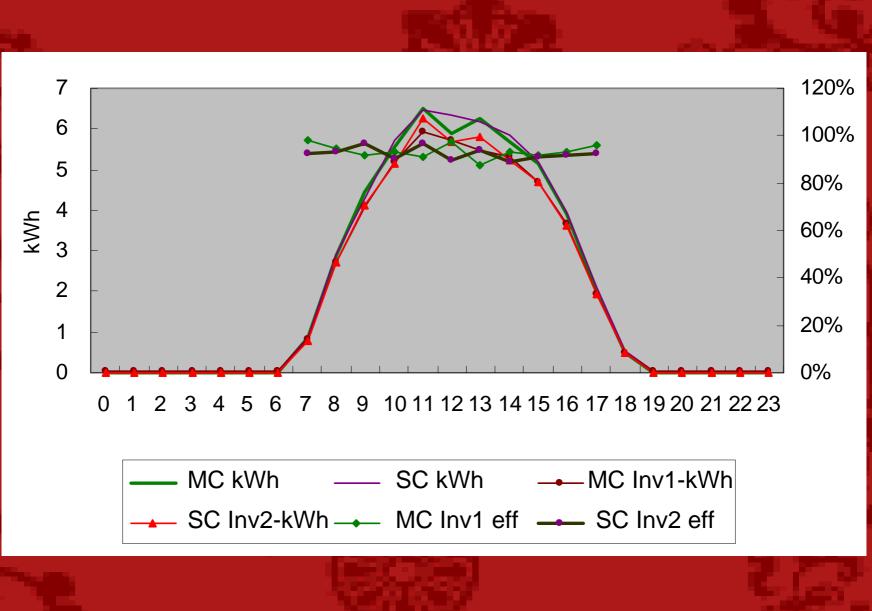
Time







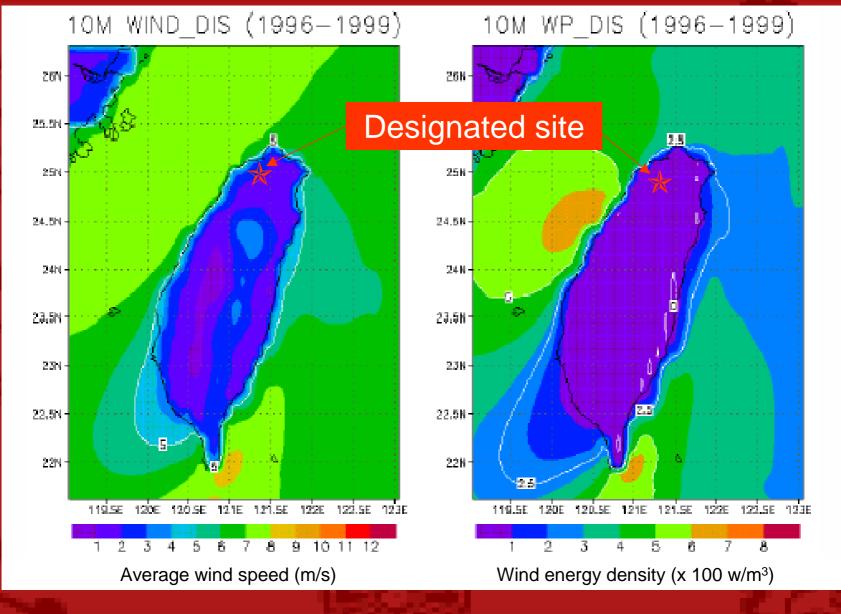




Wind Power Generation

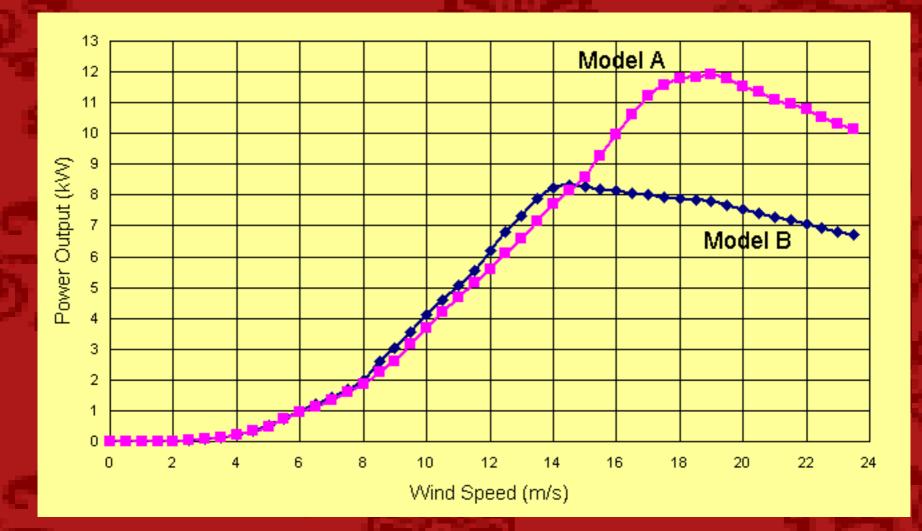






Source: Energy Commission, Ministry of Economic Affairs (http://wind.erl.itri.org.tw)







Progress

Selection of place of installation completed * A wind anemometer will be installed to assess the local wind power this year Concrete base platform will be built for both wind anemometer and turbine Conduits for power lines and data collection wires will be constructed









Remarks on Renewable Energy

The existing PV system generates a total of 15,297 kWh of electricity in a year yielding approximately 8.7 % of the total capacity Not expecting much electricity generated from the wind on site The maximum capacity of the fuel cells to be installed is 2 kW for continuous operation





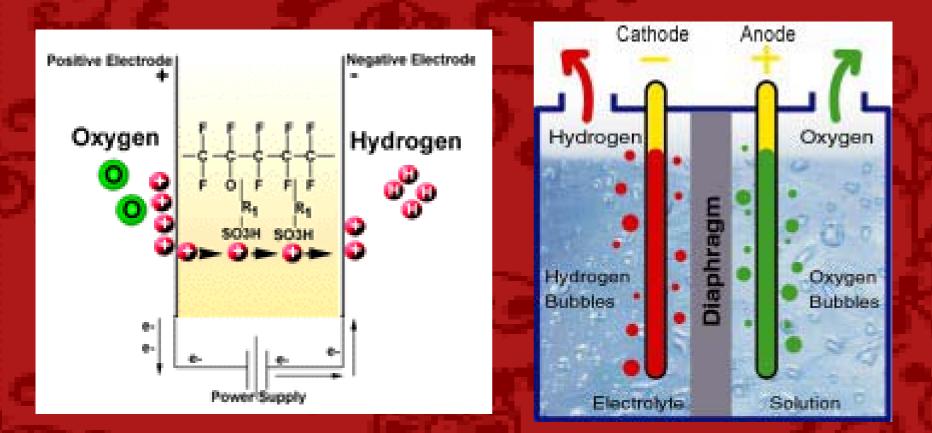
Hydrogen Production and Storage





PEM Electrolysis

Alkaline Electrolysis







General Comparison of PEM & Alkaline Water Electrolysis

Characteristic	PEM Water Electrolysis	Alkaline Water Electrolysis
Stack Configuration/Cost	 More inherently costly materials of construction, especially catalysts and membranes Capable of relatively high current densities, stack can be highly compact Generally limited output capacity per stack to date 	 Alkaline environment generally allows for use of less costly materials, especially catalysts and membranes Typical current densities are lower than for PEMWE stacks, but still quite compact. Commercial standard stack output is 15 NCMH
System Configuration/Cost	 System configuration is similar to AWE Unproven for relatively large plants (e.g., typical maximum output of 10 NCMH) 	 Commercial system configurations up to hundreds of NCMH capacity Overall system size typically is comparable to PEMWE of similar output capacity
System Reliability	 Proven for highly engineered systems in submarines Membrane is sensitive to impurities 	- Proven commercial designs, relatively robust. Over 98% availability proven in field.



SPECIFICATIONS OF A PEM WATER ELECTROLYZER

Pure hydrogen output	Up to 1.0 Nm ³ /hr			
Electrolyte system	Proton Exchange Membrane (PEM)			
	Solid electrolyte, caustic free (no KOH)	the second se		
Maximum delivery pressure	13.8 bar (200 psig)			
Hydrogen purity	99.9 + % (99.999 + % optional)			
 Hydrogen dew point 	Down to –65 °C			
Water usage	Up to 1.0 liter/hr	1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1		
Water quality requirement	Deionized ASTM Type II (minimum)			
Electrical supply required	AC: 190 – 240 VAC, 1 phase, 50/60 hz	DC: 60 – 200 VDC		
	2 kVA – 12 kVA	150 A max		
Operating environment	Indoor (outdoor optional)			
Dimensions (L x W x H)	97 cm x 105 cm x 106 cm)			
Weight	220 kg			
Shipping weight 250 kg				
Controls	Fully automatic and unattended operation			
Duty cycle	100 % (24 * 7)			
Installation "Plug & Play" expedited installation				





Comparison of Methods of Hydrogen Storage

No 5 6	Advantage	Disadvantage
System Pressure	 Low cost 	♦ Bulky
	 Simple operation 	1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1
	 Safety 	
Elevated Pressure	 Compact 	 Consume power
		 Complicated
	the beauty of	operation
Metal Hydride	Compact	 Complicated
	 Advanced 	operation
	technology	 High cost
		♦ Heavy





Metal Hydride and Hydrogen Storage

Storage	Hydrogen Stor	age Capacity	Energy	Energy Density	
Medium	Wt %	g/ml	cal/g	cal/ml	
MgH_2	7	0.101	2373	3423	
Mg ₂ NiH ₄	3.16	0.081	1071	2745	
VH_2	2.07	0.095	701	3227	
TiFeH _{1.95}	1.75	0.096	593	3254	
LaNi ₅ H ₇	1.37	0.089	464	3017	
Liquid Hydrogen	100	0.070	33900	2373	
Gaseous Hydrogen	100	0.008	33900	271	



Hydrogen Storage Capacity

△ Case 1 : 2 kW fuel cells running 24 hours

- $34.56 \text{ m}^3 \text{ (STP)} = 2.59 \text{ m}^3 \text{ (200 psig)}$
- 230 kg MnNi₅ metal hydride (< 55 liters)

△ Case 2 : 2 kW fuel cells running 48 hours

- $69.12 \text{ m}^3 \text{ (STP)} = 5.18 \text{ m}^3 \text{ (200 psig)}$
- 460 kg MnNi₅ metal hydride (< 110 liters)

△ Case 3 : 2 kW fuel cells running 72 hours

- $-103.68 \text{ m}^3 \text{ (STP)} = 7.75 \text{ m}^3 \text{ (200 psig)}$
- 690 kg MnNi₅ metal hydride (< 165 liters)

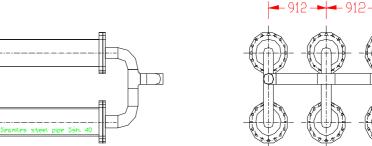
△ Case 4 : 2 kW fuel cells running 1 week

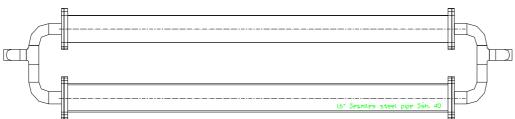
- $241.92 \text{ m}^3 \text{ (STP)} = 18.08 \text{ m}^3 \text{ (200 psig)}$
- 1,600 kg MnNi₅ metal hydride (< 385 liters)



1000





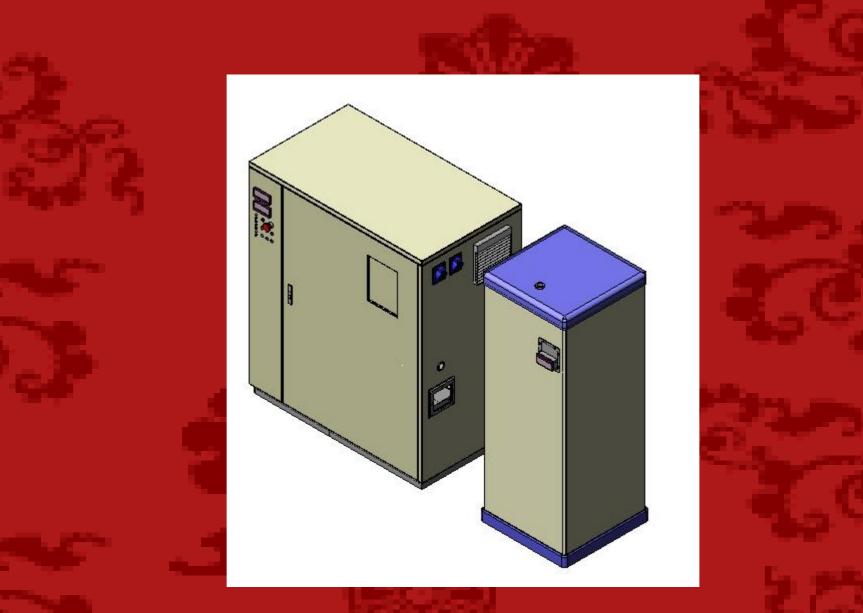


16" Seamlles steel pipe Ssh. 40 0.381*0.381*3.14159/4*6=0.684 (m3) 0.684 m3 * 12 = 0.200 m3

總重量約11000kg

Fuel Cell Power Generation

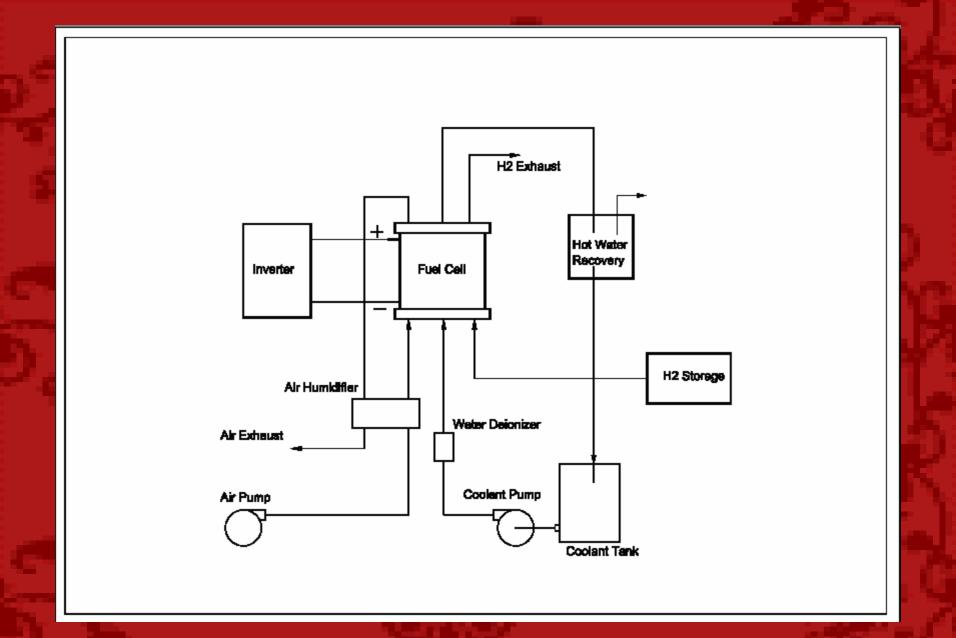




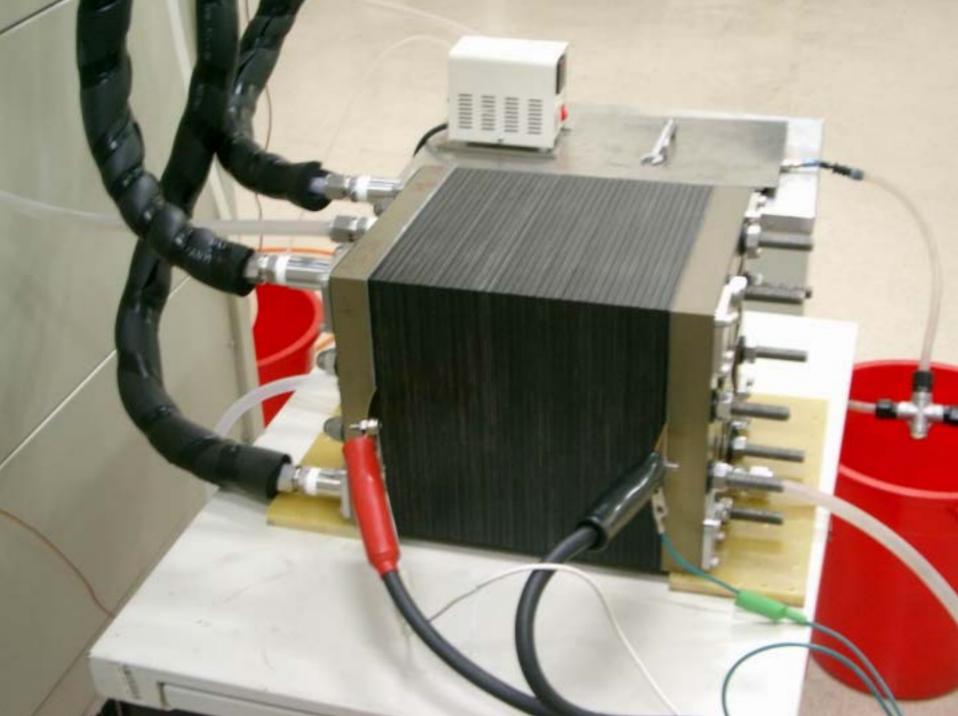
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____ Th2=65 ____ Th2=75 ____ Th2=55 Power(W) Voltage(V) Current(A)







Concluding Remarks

Hydrogen may serve as a means of storing electricity in support of distributed renewable energy based system efficiently and friendly to the environments through the use of fuel cells.

- An integrated renewable energy/fuel cell power generation system has been designed and details of the components of the system were presented.
- In addition to utilizing renewable energy more effectively, the hydrogen generator in this system could be used as a tool for load management balancing peak and off-peak electricity demand.



